

Cerro Coso College  
**Course Outline of Record Report**  
 10/07/2021

## PHSC C111 : Physical Science Lecture

### General Information

Author:	• Tech Support
Course Code (CB01) :	PHSC C111
Course Title (CB02) :	Physical Science Lecture
Department:	Science
Proposal Start:	Spring 2018
TOP Code (CB03) :	(1901.00) Physical Sciences, General
SAM Code (CB09) :	Non-Occupational
Distance Education Approved:	No
Course Control Number (CB00) :	No value
Curriculum Committee Approval Date:	10/14/2011
Board of Trustees Approval Date:	11/10/2011
External Review Approval Date:	12/15/2011
Course Description:	This course covers conceptual topics in physics and chemistry, with applications to the earth sciences and astronomy, for the non-science major. Topics such as motion, energy, electricity, magnetism, waves, atoms, chemistry and chemical reactions are covered. Not open to students who have completed PHSC 115.
Submission Type:	course integration into elumen No value
Author:	No value

### Faculty Minimum Qualifications

Master Discipline Preferred:	• Astronomy • Chemistry • Earth Science • Physical Sciences
Alternate Master Discipline Preferred:	No value
Bachelors or Associates Discipline Preferred:	No value
Additional Bachelors or Associates Discipline Preferred:	No value

### Course Development Options

Basic Skills Status (CB08) Course is not a basic skills course.	Course Special Class Status (CB13) Course is not a special class.	Grade Options • Letter Grade Methods • Pass/No Pass
	Allowed Number of Retakes	Course Prior To College Level (CB21)

Allow Students to Gain Credit by Exam/Challenge

**Allowed Number of Retakes**

0

**Course Prior to College Level (CB21)**

No value

**Rationale For Credit By Exam/Challenge**

No value

**Retake Policy Description**

Non-Repeatable Credit

Allow Students To Audit Course

**Course Support Course Status (CB26)**

No value

## Associated Programs

Course is part of a program (CB24)

**Associated Program**

**Award Type**

**Active**

CC Liberal Arts: Mathematics & Science

A.A. Degree Major

Summer 2018 to Fall 2020

CSU General Education (CSU GE Breadth)

Certificate of Achievement

Fall 2020

Intersegmental General Education Transfer Curriculum Certificate of Achievement

Certificate of Achievement

Fall 2020

Liberal Arts: Mathematics & Science Associate in Arts Degree

A.A. Degree Major

Fall 2020

CSU General Education (CSU GE Breadth) (In Development)

Certificate of Achievement

Fall 2021

Intersegmental General Education Transfer Curriculum Certificate of Achievement (In Development)

Certificate of Achievement

Fall 2021

## Transferability & Gen. Ed. Options

**Course General Education Status (CB25)**

No value

**Transferability**

Transferable to both UC and CSU

**Transferability Status**

Pending

**Cerro Coso General Education Requirements**

Categories	Status	Approval Date	Comparable Course
Natural Science Physical Sciences	Pending	No value	No Comparable Course defined.

**CSU General Education Certification**

Categories	Status	Approval Date	Comparable Course
Scientific Inquiry & Quantitative Reasoning Physical Sciences	Pending	No value	No Comparable Course defined.

**Intersegmental General Education Transfer Curriculum**

Categories	Status	Approval Date	Comparable Course
Physical & Biological Sciences Physical Science	Pending	No value	No Comparable Course defined.

**Units and Hours****Summary**

<b>Minimum Credit Units (CB07)</b>	3
<b>Maximum Credit Units (CB06)</b>	3
<b>Total Course In-Class (Contact) Hours</b>	54
<b>Total Course Out-of-Class Hours</b>	108
<b>Total Student Learning Hours</b>	162
<b>Faculty Load</b>	0

**Credit / Non-Credit Options****Course Credit Status (CB04)**

Credit - Degree Applicable

**Course Non Credit Category (CB22)**

Credit Course.

**Non-Credit Characteristic**

No Value

**Course Classification Status (CB11)**

Credit Course.

**Funding Agency Category (CB23)**

No value

 Cooperative Work Experience Education Status (CB10)
 Variable Credit Course**Weekly Student Hours**

	In Class	Out of Class
Lecture Hours	3	6

**Course Student Hours**

<b>Course Duration (Weeks)</b>	18
<b>Hours per unit divisor</b>	54

Laboratory Hours	0	0
Activity Hours	0	0

**Course In-Class (Contact) Hours**

Lecture	54
Laboratory	0
Activity	0
<b>Total</b>	<b>54</b>

**Course Out-of-Class Hours**

Lecture	108
Laboratory	0
Activity	0
<b>Total</b>	<b>108</b>

**Time Commitment Notes for Students**

No value

**Faculty Load**

Extra Duties: 0

Faculty Load: 0

**Units and Hours - Weekly Specialty Hours**

Activity Name	Type	In Class	Out of Class
No Value	No Value	No Value	No Value

**Pre-requisites, Co-requisites, Anti-requisites and Advisories**

**Prerequisite**

MATHC055 - Intermediate Algebra

Students entering PHSC C111 are required to solve problems involving mathematical operations such as ratios, square roots, surface areas related to radius, and solving for a single variable.

**AND**

**Advisory**

ENGLC070 - Introductory Composition

Reading - 1 Level Prior to Transfer  
Content Review

Students entering PHSC C111 are expected to identify central points, both explicit and implied, of scientific periodical articles and textbooks, outline and summarize complex and technical scientific readings, interpret difficult and figurative language: academic discourse and scientific terminology, write lab reports in an accepted format. Students are also expected to answer essay questions in clear and error free prose based on readings from texts and scientific journals and also outline and summarize assigned readings texts and scientific journals.

The reading advisory level provides the student with the requisite skills to meet this expectation.

Writing - 2 Levels Prior to Transfer

Content Review

Students entering PHSC C111 are expected to identify central points, both explicit and implied, of scientific periodical articles and textbooks, outline and summarize complex and technical scientific readings, interpret difficult and figurative language: academic discourse and scientific terminology, write lab reports in an accepted format. Students are also expected to answer essay questions in clear and error free prose based on readings from texts and scientific journals and also outline and summarize assigned readings texts and scientific journals.

The writing advisory level provides the student with the requisite skills to meet this expectation.

## Entrance Skills

Entrance Skills	Description
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No value	No value
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## Limitations on Enrollment

Limitations on Enrollment	Description
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No value	No value
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## Specifications

Methods of Instruction	
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Methods of Instruction	Demonstration
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Rationale	No value
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Methods of Instruction	Discussion
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Rationale	No value
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Methods of Instruction	In-class writing
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Rationale	No value
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Methods of Instruction	Lecture
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Rationale	No value
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<b>Methods of Instruction</b>	Outside reading
<b>Rationale</b>	No value
<b>Methods of Instruction</b>	Peer analysis, critique & feedback
<b>Rationale</b>	No value
<b>Methods of Instruction</b>	Presentations (by students)
<b>Rationale</b>	No value
<b>Methods of Instruction</b>	Problem Solving
<b>Rationale</b>	No value
<b>Methods of Instruction</b>	Written work
<b>Rationale</b>	No value
<b>Methods of Instruction</b>	Other
<b>Rationale</b>	Recitation
<b>Assignments</b>	
<p>A. Homework assignments from the relevant chapter, including participation in the recitation/discussion session. Example: Students must solve problems on distance, velocity and acceleration and participate in discussions distinguishing the relationship among these three terms. B. Assigned readings from the textbook and/or other sources. Example: Students must read the relevant chapter on forces and how to analyze the forces that a person parachuting to earth encounters. C. Research Paper/Presentation. Example: Students are required to present a paper on the chemicals found in their household and the benefits and dangers of each. D. Critical Analysis of course relevant topics that appear in the media. Example: Students are required to read two science-based (not popular media) articles on both sides of the climate change discussion and present a cogent synopsis, including the strengths and weaknesses of each paper.</p>	
<b>Methods of Evaluation</b>	<b>Rationale</b>
Tests	Exams and Quizzes evaluate the students' ability to apply techniques taught in class and apply these techniques in problem solving. Example: The first midterm exam requires students to conceptually solve equations relating to the motion of an object under the influence of gravity and air resistance.
Homework	Regular homework assignments reinforce material learned in class and evaluate the student's ability to learn outside the classroom. Example: A homework assignment covers the conceptual analysis of electric current in the household
Participation	

Participation in Problem Solving evaluates the student's ability to solve problems in a group environment. Example: Students participate in the analysis of the speed of sound

#### Equipment

No Value

#### Textbooks

Author	Title	Publisher	Date	ISBN
Hewitt, P. G. Suchocki, J., & Hewitt, L. A. .	Conceptual Physical Science. , 5th,	Addison-Wesley	2012	

#### Other Instructional Materials

No Value

#### Materials Fee

No value

### Learning Outcomes and Objectives

#### Course Objectives

No value

#### CSLOs

Solve problems related to motion, momentum and energy using the appropriate theoretical concept. Expected SLO Performance: 70.0

Perform an analysis of thermodynamic concepts in order to solve problems in heat transfer and phase change. Expected SLO Performance: 70.0

Use the concepts of electromagnetism and wave theory to solve problems related to electricity, magnetism, waves and sound. Expected SLO Performance: 70.0

Solve problems involved in basic chemistry, chemical bonding, reactions and mixtures using the concepts of conservation of mass, bonding theory and atomic theory. Expected SLO Performance: 70.0

Analyze and utilize the scientific method and proper scientific formatting in problem solving. Expected SLO Performance: 70.0

### Outline

Course Outline

- A. Motion and Equilibrium
  - 1 Aristotle On Motion
  - 2 Galileo's Concept of Inertia
  - 3 Mass—A Measure of Inertia
  - 4 Net Force
  - 5 The Equilibrium Rule
  - 6 Support Force
  - 7 Equilibrium Of Moving Things
  - 24.
  - 7 Equilibrium Of Moving Things
  - 8 The Force of Friction
  - 9 Speed and Velocity
    - a. Speed
    - b. Instantaneous speed
    - c. Average speed
    - d. Velocity
  - 10 Acceleration
- B. Newton's Laws of Motion
  - 1. Newton's First Law Of Motion
  - 2. Newton's Second Law of Motion
  - 3. Forces and Interactions
  - 4. Newton's Third Law of Motion
  - 5. Vectors
  - 6. Summary of Newton's Three Laws
- C. Momentum and Energy
  - 1. Momentum
  - 2. Impulse
  - 3. Impulse-Momentum Relationship
  - 4. Conservation of Momentum
    - a. Collisions
  - 5. Energy
    - a. Work
    - 6. Power
    - 7. Potential Energy
    - 8. Kinetic Energy
      - a. Work-Energy Theorem
      - b. Kinetic Energy and Momentum Compared
  - 9. Conservation of Energy
  - 10. Machines
    - a. Efficiency
- D. Thermal Energy and Thermodynamics
  - 1. Thermal Energy
  - 2. Temperature
  - 3. Absolute Zero
  - 4. Heat
  - 5. Quantity of Heat
  - 6. The Laws of Thermodynamics
  - 7. Specific Heat Capacity
  - 8. Thermal Expansion
    - a. Expansion of Water
- E. Heat Transfer and Change of Phase
  - 1. Conduction
  - 2. Convection
  - 3. Radiation
    - a. Emission of Radiant Energy
    - b. Absorption of Radiant Energy
    - c. Reflection of Radiant Energy
    - d. Cooling at Night by Radiation
  - 4. Newton's Law Of Cooling
  - 5. Heat Transfer and Change of Phase
  - 6. Evaporation
  - 7. Condensation
  - 8. Boiling
  - 9. Melting and Freezing
  - 10. Energy and Change of Phase
- F. Static and Current Electricity
  - 1. Electric Force and Charge



- a. Conservation of Charge
- 2. Coulomb's Law
- a. Charge Polarization
- 3. Electric Field
- 4. Electric Potential
- 5. Voltage Sources
- 6. Electric Current
- a. Direct Current and Alternating Current
- 7. Electrical Resistance
- 8. Ohm's Law
- a. Electric Shock
- 9. Electric Circuits
- a. Series Circuits
- b. Parallel Circuits
- c. Parallel Circuits and Overloading
- d. Safety Fuses
- 10. Electric Power
- G. Magnetism and Electromagnetic Induction
- 1. Magnetic Poles
- 2. Magnetic Fields
- 3. Magnetic Domains
- 4. Electric Currents and Magnetic Fields
- a. Electromagnets
- b. Superconducting Electromagnets
- 5. Magnetic Forces on Moving Charges
- a. Magnetic Force on Current-Carrying Wires
- b. Electric Meters
- c. Electric Motors
- 6. Electromagnetic Induction
- a. Faraday's Law
- 7. Generators and Alternating Current
- 8. Power Production
- 9. The Transformer—Boosting or Lowering Voltage
- 10. Field Induction
- H. Waves and Sound
- 1. Vibrations and Waves
- 2. Wave Motion
- a. Wave Speed
- 3. Transverse and Longitudinal Waves
- 4. Sound Waves
- a. Speed of Sound
- 5. Reflection of Sound
- 6. Refraction of Sound
- 7. Forced Vibrations
- 8. Resonance
- 9. Interference
- a. Beats
- b. Standing Waves
- 10. Doppler Effect
- 11. Wave Barriers And Bow Waves
- 12. Shock Waves and the Sonic Boom
- 13. Musical Sounds
- a. Musical Instruments
- I. Light Waves
- 1. Electromagnetic Spectrum
- 2. Transparent and Opaque Materials
- 3. Color
- a. Selective Reflection
- b. Selective Transmission
- c. Mixing Colored Lights
- d. Mixing Colored Pigments
- e. Why the Sky Is Blue
- f. Why Sunsets Are Red
- g. Why Clouds Are White
- 4. Diffraction
- 5. Interference
- a. Interference Colors by Reflection from Thin Films

- 6. Polarization
- J. Atoms and Periodic Table
  - 1. The Elements
  - 2. Atoms Are Ancient and Empty
  - 3. Protons and Neutrons
  - 4. Isotopes and Atomic Mass
    - a. Figuring Physical Science: Calculating Atomic Mass
  - 5. The Periodic Table
  - 6. Periods and Groups
- K. Elements of Chemistry
  - 1. Chemistry: The Central Science
  - 2. The Submicroscopic World
  - 3. Physical and Chemical Properties
  - 4. Determining Physical and Chemical Changes
  - 5. Elements to Compounds
  - 6. Naming Compounds
  - 7. Chemical Equations
    - a. Balancing Unbalanced Equations
- L. Mixtures
  - 1. Most Materials Are Mixtures
    - a. Mixtures Can Be Separated By Physical Means
  - 2. The Chemist's Classification of Matter
  - 3. Solutions
  - 4. Purifying the Water We Drink
  - 5. Desalination
  - 6. Wastewater Treatment
    - a. Advanced Integrated Pond Systems
- M. How Atoms Bond
  - 1. Electron-Dot Structures
  - 2. The Formation of Ions
    - a. Molecules Can Form Ions
  - 3. Ionic Bonds
  - 4. Covalent Bonds
  - 5. Polar Covalent Bonds
  - 6. Molecular Polarity
  - 7. Metallic Bonds
    - a. We Should Conserve and Recycle Metals
- N. Chemical Reaction
  - 1. Reaction Rates
  - 2. Catalysts
  - 3. Energy and Chemical Reactions
    - a. An Exothermic Reaction Involves a Net Release of Energy
    - b. An Endothermic Reaction Involves a Net Absorption of Energy
  - 4. Relative Masses of Atoms and Molecules
  - 5. Molar Mass